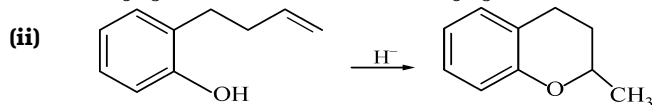
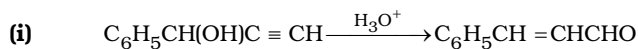
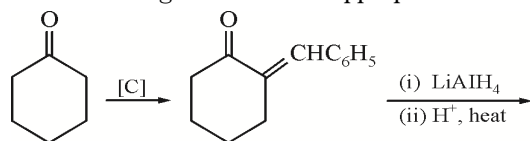


Date Planned : __ / __ / __	Daily Tutorial Sheet-4	Expected Duration : 45 Min
Actual Date of Attempt : __ / __ / __	JEE Advanced (Archive)	Exact Duration : _____

45. Write the intermediate steps for each of the following reactions (1998)



46. Complete the following reaction with appropriate structure of products/reagents. (1998)

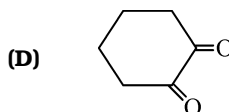
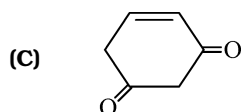
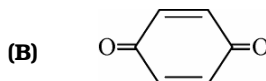
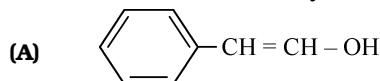


47. An aldehyde  $\text{A}(\text{C}_{11}\text{H}_{18}\text{O})$ , which does not undergo self aldol condensation, gives benzaldehyde and two moles of B on ozonolysis. Compound B, on oxidation with silver ion gives oxalic acid. Identify the compounds A and B. (1998)

48. Which of the following will react with water ? (1998)

- (A)  $\text{CHCl}_3$       (B)  $\text{Cl}_3\text{CCHO}$       (C)  $\text{CCl}_4$       (D)  $\text{ClCH}_2\text{CH}_2\text{Cl}$

\*49. Tautomerism is exhibited by : (1998)



\*50. A new carbon-carbon bond formation is possible in : (1998)

- (A) Cannizzaro's reaction      (B) Friedel-Crafts' reaction  
(C) Clemmensen's reduction      (D) Reimer-Tiemann reaction

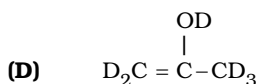
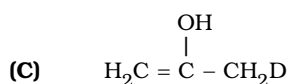
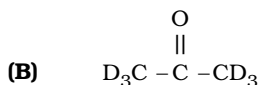
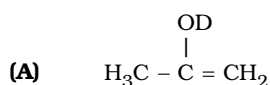
\*51. Which of the following will undergo aldol condensation ? (1998)

- (A) Acetaldehyde      (B) Propanaldehyde  
(C) Benzaldehyde      (D) Trideutero acetaldehyde

\*52. Among the following compounds, which will react with acetone to give a product containing  $\text{>C=N-}$  ? (1998)

- (A)  $\text{C}_6\text{H}_5\text{NH}_2$       (B)  $(\text{CH}_3)_3\text{N}$       (C)  $\text{C}_6\text{H}_5\text{NHC}_6\text{H}_5$       (D)  $\text{C}_6\text{H}_5\text{NHNH}_2$

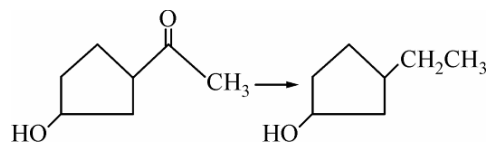
53. The enol form of acetone, after treatment with  $\text{D}_2\text{O}$ , gives : (1999)



54. (a) Compound A ( $C_8H_8O$ ) on treatment with  $NH_2OH$ ,  $HCl$  gives B and C. B and C rearrange to give D and E, respectively, on treatment with acid. B, C, D and E are all isomers of molecular formula ( $C_6H_9NO$ ). When D is boiled with alcoholic  $KOH$ , an oil F ( $C_6H_7N$ ) separates out. F reacts rapidly with  $CH_3COCl$  to give back D. On the other hand, E on boiling with alkali followed by acidification gives a white solid G ( $C_7H_6O_2$ ). G Identify A-G. (1999)

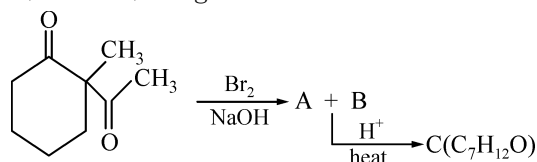
(b) Carry out the following transformation in not more than three steps. 1-butyne  $\longrightarrow$  2-pentanone

55. The appropriate reagent for the following transformation : (2000)



- (A)  $Zn(Hg), HCl$  (B)  $NH_2NH_2, OH^-$  (C)  $H_2 / Ni$  (D)  $NaBH_4$

56. Identify A, B and C, and give their structures. (2000)



57. An organic compound A,  $C_6H_{10}O$ , on reaction with  $CH_3MgBr$  followed by acid treatment gives compound B. The compound B on ozonolysis gives compound C, which in presence of a base gives 1-acetylcyclopentene D. The compound B on reaction with  $HBr$  gives compound E. Write the structures of A, B, C and E. Show, how D is formed from C. (2000)

58. What would be the major product in the following reaction ? (2000)

